

**EFFECT OF DIFFERENT DRYING TEMPERATURES ON THE
COMPOSITION AND ANTIOXIDANT ACTIVITY OF GINGER POWDER**

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The influence of drying temperatures ranging from 40° to 100°C on the chemical composition, antioxidant properties and microstructure characteristics of the ginger powders was investigated. The ginger samples dried at 100°C showed a significantly higher amount of polyphenols (24.154 mg EAG/g) and flavonoids (10.564 mg EAG/g) contents. Moreover, the antioxidant activity increased from 73.47% at 40°C to 78.23% at 100°C. The reduction trend of 6-gingerol and β -carotene concentrations was obtained by rising the drying temperatures as indicated by high performance liquid chromatography (HPLC) analysis. In contrast, zingerone and 6-shogol contents significantly increased at high drying temperatures. A pronounced gelatinization and a more compacted structure was observed in the ginger powders dried at high temperatures (80 and 100°C) as indicated by scanning electron microscopy analysis. These findings offer a better comprehension of the influence of the oven drying process on the functional properties and structure characteristics of the ginger powder, hence allowing the optimization and development of applications in the food and pharmaceutical industries.

Keywords: dried ginger, antioxidant activity, flavonoids, 6-gingerol, 6-shogol, zingerone

Introduction

Belonging to the family Zingiberaceae, Ginger (*Zingiber officinale Roscoe*) is one of the most common plants globally used in food and beverages as a spice and flavouring agent. Ginger has been used for more than 2000 years in many cultures (Bartley and Jacobs, 2000). For instance, Asians, Egyptians, Greeks and Romans

were well aware of the ginger's nutritional and medical importance and often used it for the treatment of various symptoms such as nausea, stomach discomfort, diarrhoea, headache and common cold (Miyoshi *et al.*, 2003). Various studies demonstrated the strong pharmacological properties of ginger as an anti-diabetic, anti-inflammatory (Minghetty *et al.*, 2007), anti-microbial, anti-carcinogenic, analgesic (Li *et al.*, 2012), anti-vomiting agent and to decrease the intensity of pregnancy-induced nausea (Ozgoli *et al.*, 2009). It is known that ginger contains numerous of pharmacologically active compounds. They are essentially presented as volatiles compounds (essential oils), gingerols and shogaols (Li *et al.*, 2016). Gingerols are a series of homologous phenols; essentially 6, 8, and 10 gingerol, contributing to the ginger's pungency. Shogaols are the dehydrated form of gingerols and gingerdiones mainly 6, 8, and 10 shogaol and zingerone (Jiang *et al.*, 2005). In general, ginger rhizomes are commercially sold fresh. With the benefits of practical handling and long shelf life, ginger powder has become an essential solution for the non-producing countries such as Algeria, as a food and/or pharmaceutical ingredient. Drying is one of the conventional methods commonly used to preserve food due to its simplicity and low cost. It is known that dehydration or drying process was applied in foods in order to extend their shelf-life by reducing the water activity (Prasad *et al.*, 2006), which is essential for the growth of bacteria, mould and yeasts. In the food drying process, temperature and process duration are key factors for the product's preservation quality and energy consumption. Various comparative studies were investigated regarding different drying processes such as sun drying, oven drying and microwave drying (Cheng *et al.*, 2011; Li *et al.*, 2016). However, few studies focused on finding the optimal temperature applied in the same drying process that allows both the obtainment of a high quality ginger powder and also the protection of bioactive compounds with maximum beneficial properties.

The objective of this work was to evaluate the effect of various drying temperatures ranging from 40° to 100°C on the quality attributes of dried ginger powder. Total phenolic, flavonoid compounds, carotenoids, total protein, fat contents and the antioxidant activity of the ginger powders were determined. Furthermore, the different concentrations of 6- gingerol, 6-shogaol, zingerone and β -carotene were assessed by HPLC. Scanning electron microscopy (SEM) was used to examine the structural characteristics of the obtained ginger powders.

Materials and Methods

Reagents and chemicals

Sodium carbonate, 2, 2-diphenyl-1-picryl-hydrazyl (DPPH), Folin–Ciocalteu reagent (FCR), standard chemicals (zingerone, 6-shogaol, 6-gingerol and β -carotene), gallic acid, catechin, sodium hydroxide, aluminium chloride, sodium nitrate were purchased from Sigma-Aldrich (Dublin, Ireland). Analytical grade acetone, acetonitrile, ethanol, ethyl acetate, methanol and hexane were attained from Merck (Gernsheim, Germany).

Plant material and sample preparation

Fresh mature ginger rhizomes (origin from China) were purchased during the time between March and June, from a local Algerian market (Bejaia city). The rhizomes were thoroughly washed and rinsed to remove impurities. The washed ginger rhizomes were then chopped into 3 mm thick slices. The average moisture content of the ginger samples was determined using the Association of Official Analytical Chemist method (AOAC 2000). Briefly, individual loads of 100 g of chopped ginger slices were subjected to oven (Binder E28, Germany) drying at 40°, 60°, 80° and 100°C until a constant weight was obtained. The mass was periodically measured during the drying process and the weight loss (%) was calculated as follows:

$$\text{Weight loss (\%)} = \frac{W_t \times 100}{W_0} \quad (1)$$

where, W_0 and W_t (g) are the initial weight and the weight at a specific time, respectively.

Once dried, the ginger slices were ground into powders using an electric grinder (IKA model-A11, Baden-Württemberg, Germany) to a granulometry lower than 250 μm in size. Finally, the sieved ginger powders were individually transferred to air-tight amber glass containers and stored in the dark until further use.

Extraction of phenolic compounds

The phenolic contents were extracted from each individual ginger powder in accordance to a method previously developed by Mukherjee *et al.* (2012). Briefly, 1 g of individual ginger powder was extracted at 40°C under continuous agitation using 50 ml of 5% (v/v) ethanol. The extract mixture was then centrifuged at 1,800 $\times g$ for 30 min, followed by filtration using Whatman qualitative filter paper No. 1. The filtrates were afterwards transferred to air-tight amber glass containers and stored at 4°C until further analysis.

Determination of total phenolic content

Total phenolic content (TP) from the ginger's ethanolic extracts was measured by the Folin-Ciocalteu assay as described by Chan *et al.* (2008), with some minor modifications. Two hundred μl of each ginger extract (prepared in section 2.3) were mixed with 500 μl of 1:10 (v/v) aqueous Folin-Ciocalteu reagent and 1500 μL of sodium bicarbonate (75 g/l), vortexed, then incubated for 30 minutes at room temperature. The absorbance was recorded at 765 nm using an ultraviolet (UV) spectrophotometer (UV-mini 1240, Shimadzu, Japan). The TP content of the ginger extracts was calculated from the calibration curve using gallic acid as the standard. The samples were analysed in triplicate and the mean value of three tests was used. The results were expressed as milligram of gallic acid equivalent per gram of dry sample (GAE/g).

Determination of total flavonoids content

The total flavonoids content (TF) can be determined by quantifying the yellow colour intensity developed by the complex flavonoids- Al^{3+} formed due to the chelating of the hydroxyls group (OH^{-1}) in flavonoids. According to the method

described by Jia *et al.* (1999), 500 µl of ginger extracts (prepared in section 2.3) were diluted with 2 ml of distilled water, then 150 µl of 5% (w/v) NaNO₂ solution was added. This mixture was allowed to stand for 6 minutes. Afterwards, 150 µl of 10% (w/v) AlCl₃ solution was added to the mixture and allowed to stand for another 6 minutes. Finally, 2 ml of 4% (w/v) NaOH solution were added, mixed and incubated further in the darkness for 15 minutes at room temperature. The absorbance was measured at 510 nm using an ultraviolet (UV) spectrophotometer (UV-mini 1240, Shimadzu, Japan). The samples were independently analysed in triplicate and the mean value of three tests was used. The results were expressed as milligram of catechin equivalent per gram of dry weight (mg CE/g) from the standard curve prepared with catechin.

Determination of total carotenoids content

The total carotenoid content (TC) was measured according to Sass-Kiss *et al.* (2005). A hundred mg of dried ginger powder was accurately weighed and suspended for 30 minutes in 10 ml of 2:1:1 (v/v) mixture of hexane, acetone and ethanol, respectively. This mixture was vortexed for 15 minutes, centrifuged at 2250 × g and the top layer was then recovered. The absorbance of the separated hexane portion was measured at 450 nm using an ultraviolet (UV) spectrophotometer (UV-mini 1240, Shimadzu, Japan). The extracted samples were independently analysed in triplicate and the mean value of three measurements was used. The results were calculated on the basis of the calibration curve that was obtained by using β-carotene as the standard and expressed as milligram of β-carotene equivalent per 100 gram of dry ginger (mg βCE/100 g).

Determination of the DPPH radical scavenging activity

The antioxidant activity was evaluated by the DPPH radical scavenging assay according to the method described by Brand-Williams *et al.* (1995) with minor modifications. Briefly, 3ml of 70 µM DPPH already dissolved in methanol were mixed with 100 µl of each ginger extract (prepared in section 2.3). After incubation at room temperature for 30 minutes at 37°C, the absorbance was measured at 517 nm using an ultraviolet (UV) spectrophotometer (UV-mini 1240, Shimadzu, Japan). The results were expressed as the percentage of inhibition of the DPPH radical and calculated as follows:

$$\text{DPPH inhibition \%} = \left(\frac{A_c - A_s}{A_c} \right) \times 100 \quad (2)$$

where: A_c and A_s are the absorbance of the control and the sample, respectively. The samples were analysed in triplicate.

Determination of 6-gingerol, 6-shaogaol and zingerone by HPLC

HPLC analysis was performed using an Agilent 1260 Infinity Series system (Agilent Technologies, Waldbronn, Germany) equipped with a G1311B 1260 quaternary pump, G1329B 1260 autosampler, vacuum degasser, G1316A 1260 temperature-controlled column compartment and G1315D 1260 DAD VL detector. Standard solutions of zingerone, 6-gingerol and 6-shogaol were individually prepared in the range of 2 µg/ml to 500 µg/ml. The column used was Kromasil100

C18 (250 × 4.6 mm, 5µm). The column temperature was maintained at 40°C. The typical injection volume was 10 µl. The eluents were 0.1% (v/v) phosphate buffer (solvent A) and acetonitrile (solvent B). The gradient elution profile was 0-3 min: 30- 70% A; 3- 40 min: 30- 70% B, then at 40 - 50 min 30- 70% A. The mobile phase flow rate was adjusted to 1.0 ml/min. The detection wavelength was set at 280 nm. HPLC data was collected and analysed by Agilent OpenLAB 2.2 chromatography data systems (CDS) software. The identification of zingerone, 6-gingerol and 6-shogaol components in the ethanol ginger extracts (see section 2.3) was achieved by comparing their retention times (RT) to those of the used standards.

Determination of β-carotene by HPLC

The β-carotene quantification in ginger was determined by HPLC analysis as reported by Stinco *et al.* (2014) and Benmeziane *et al.* (2018). Briefly, 10 mg of ginger powder were mixed with 1 ml of Milli-Q-water. The sample mixtures were centrifuged for 3 minutes at 1,800 ×g and the supernatant was discarded. One millilitre of the 1:1 (v/v) acetone: hexane extraction solvent was then added to each sample, centrifuged, vortexed and re-centrifuged for 3 minutes at 1,800 ×g. The supernatant was then collected. These steps were repeated until the yellow coloration disappeared. The organic fraction was then evaporated under N₂. The obtained dried residues were re-dissolved in 100 µl of acetonitrile for the HPLC assay. The standard solutions of β-carotene were prepared in the range between 0.005 µg/ml and 2.8 µg/ml. The separation was carried out using a Poroshell 120 EC-C18 column (4.6× 50 mm, 2.7 µm). The column temperature was maintained at 40°C. The used eluents were ethyl acetate (solvent A), methanol (solvent B) and acetonitrile (solvent C). The linear gradient elution settings were: 0-5 min: 85% A, 15% B; 5-9 min: 60% A, 20% B, 20% C; 9-15 min, 85% A, 15% B. The mobile phase flow rate was adjusted to 1.0 mL/min. The detection wavelength was set at 280 nm. The HPLC data was collected and analysed by Agilent OpenLAB 2.2 chromatography data systems (CDS) software. The β-carotene identification was achieved by comparing their retention times (RT) to those of the used standards.

Determination of moisture, fat and protein contents

Moisture and total solids (TS) components of the ginger powders were determined by the SMART-6TM Analyzer (CEM Corporation, NC, USA). The fat content was quantified using the ORACLE Universal Fat Analyzer (CEM Corporation, NC, USA) that utilizes its proprietary nuclear magnetic resonance (NMR) signal to accurately measure the fat content of any product without the need for method development. Total protein and water-soluble protein (WSP) components were determined using the SprintTM Rapid Protein Analyzer (CEM Corporation, NC, USA), previously standardized for crude protein determination according to Kjeldahl method (AOAC, 2000). The tests were performed in triplicate.

Water-soluble protein extraction

Two grams of ginger powder were weighed accurately and suspended in 100 mL Milli-Q water. The samples were shaken overnight (12-15 hours) at room

temperature using the LABOSHAKE shaker (Gerhardt GmbH, 53639 Königswinter, Germany), centrifuged at $2,500 \times g$ for 20 min, then filtered using the Whatman No.1 filter paper. The supernatant was used for protein determination (WSP %) as described in section 2.10. Each ginger powder sample was extracted in triplicate.

Microstructure of the ginger powder

The differences of the microstructure characteristics of the ginger powders as a function of different drying temperatures were analysed using a Hitachi S700 scanning electron microscope (SEM) (Hitachi, Berkshire SL6 8YA, England). All the samples were scattered and mounted on SEM aluminium stubs, sputter-coated with gold under vacuum (EMITECH-K550, Ashford, England). The SEM images were obtained by examining the powder particles at 1.5 KV at a magnification of 1000 x.

Statistical Analysis

The analysis of variance (ANOVA) was performed with Minitab® 18.1 (MINITAB Inc., Coventry, UK), using the general linear model (GLM) procedure and the Tukey's test for pairwise comparison. The Pearson correlation analysis was used to explore the relationship between variables. All measurements were carried out in triplicate and the results were expressed as the mean value \pm standard deviation (SD).

Results and Discussion

Drying kinetic and moisture loss

The ginger slices were dried from an initial moisture content of 90% to less than 10% using different temperatures. The drying kinetics and the moisture content profiles are presented in Figure 1.

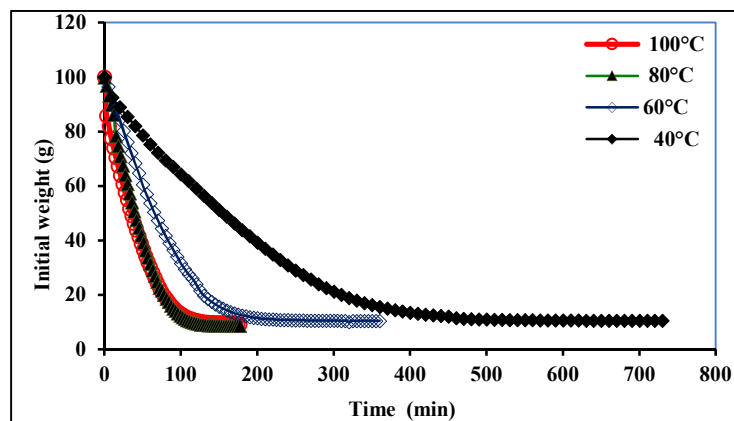


Figure 1. Drying kinetics and moisture loss of the ginger samples obtained at different drying temperatures

As expected, the drying times were inversely related to the applied drying temperatures. Drying at 40°C showed the lowest moisture loss rate, the drying process being accomplished after 710 min.

As it can be seen from the Table 1, the drying time was reduced by more than 50% when the drying temperature increased from 40°C to 60°C. These findings are in agreement with those of Arslan and Ozcan (2011), who reported a faster water loss at high drying temperatures. As shown in Fig. 1, the drying process was fast in the initial phase due to the high water content, which led to a high moisture evaporation rate. Therefore, the major drying process occurred in the range of the falling rate period, which resulted from the predominance of the internal diffusion mechanism due to the presence of bound-water (Karathanos and Belessiotis, 1990; Kaymak-Ertekin, 2002).

Table 1. Drying time and final weight at the end of the ginger powders drying process

Drying temperature, °C	Final ginger weight*, %	Drying time, min
40	10.58 ± 0.77	710 ± 15.1
60	10.48 ± 0.68	340 ± 11.0
80	9.69 ± 0.52	177 ± 7.5
100	9.42 ± 0.46	170 ± 6.6

Values are the means of triplicate analysis ± standard deviation.

* Final ginger weight was recorded at a constant interval time (min) until achieving a constant weight, which is expressed as percentage.

In this context, Baker (1997) reported that the water in larger capillaries was removed first, followed by that in smaller capillaries with a decrease of the evaporation rate. By approaching the end of the drying process, the drying rate decreases so that water becomes more difficult to be removed due to the high bound-water to the sites of the food components; i.e. protein and starch.

Total phenolic, flavonoid and carotenoids compounds

Phenolic content

Total phenolic contents (TP) of different dried ginger samples, expressed as mg of gallic acid per gram of dry sample, are given in Table 2. TP values in the ginger extracts were significantly different ($P < 0.05$) and were positively related to the drying temperatures. The ginger sample dried at 100°C exhibited the highest TP value (24.15 mg GAE/g) compared to the samples dried at lower temperatures; i.e. 16.36, 14.48 and 12.21 mg GAE/g for the samples dried at 80°, 60° and 40°C, respectively. It is known that thermal processing is the most critical factor that may influence the amount of the phenolics. Kubra and Rao (2012) reported that heat energy may cause the breakdown of the cellular constituents leading to a higher release of polyphenols from the matrices, which may explain the high content of TP obtained at 100°C in the current study. On the other hand, Jelled *et al.* (2015) concluded that a decline trend of the TP values could be attributed to the degradation of phenolic compounds during the drying process, powdering and

storage. Different values of phenolic compounds have been reported by other studies on dried ginger. For instance, An *et al.* (2016) and Gumusay *et al.* (2015) have reported values of 9.69 mg GAE/g and 3.54 mg GAE/g, respectively, for the ginger powder dried at 60°C. Whereas, Li *et al.* (2016) reported a content of 27.40 mg GAE/g for the dried ginger obtained at 40°C. The differences among these results could be attributed to the variety, genetics and the origin of the ginger rhizomes used.

Table 2. Effect of drying temperatures on total phenolic, flavonoid, carotenoid content and the antioxidant activity of ginger powder

Treatment	Total phenolics, mg/g	Total flavonoids, mg/g	Total carotenoids, mg /100g	% Inhibition DPPH
40°C	12.21 ± 0.246 ^d	2.70 ± 0.167 ^d	9.26 ± 0.092 ^a	73.47 ± 0.376 ^d
60°C	14.48 ± 0.793 ^c	3.76 ± 0.383 ^c	6.90 ± 0.132 ^b	74.95 ± 0.512 ^c
80°C	16.36 ± 0.95 ^b	5.98 ± 0.221 ^b	5.78 ± 0.092 ^c	77.04 ± 0.256 ^b
100°C	24.15 ± 0.65 ^a	10.57 ± 0.383 ^a	3.06 ± 0.134 ^d	78.23 ± 0.246 ^a

The values presented are the means ± standard deviation (n=3). The means in the same column bearing a different superscript letter differs significantly ($P < 0.05$).

Flavonoid content

The total flavonoid content (TF) exhibited the same trend as TP. As it can be seen from the data in Table 2, the TF values significantly ($P < 0.05$) increased by rising the drying temperatures. The highest amount of TF (10.57 mg CE/g) was noticed in the ginger powder dried at the high temperature of 100°C. This amount decreased to 5.98 mg for the ginger dried at 80°C, and a further TF loss was obtained by decreasing the drying temperature so that at 40°C a 2.7 mg CE/g value was reached. Our results are in accordance o those of An *et al.* (2016) and Shieber *et al.* (2001) who reported that a long period of time and a high temperature during the drying process are among the crucial factors that influence the flavonoids content. High drying temperatures contribute to the destruction of cellular constituents, which release the flavonoids and make them available during extraction. On the other hand, a short exposure time during the drying process makes it advantageous in terms of flavonoid preservation.

Carotenoid content

The total carotenoids (TC) were assessed in different ginger powders and the results are illustrated in Table 2. The results showed significant differences ($P < 0.05$) for the (TC) contents in the ginger powders with an inverse bearing of the drying temperature. The highest (TC) content (9.26 mg/100 g) was recorded for the ginger dried at 40°C. This value steadily decreased by increasing the drying temperature to reach a TC concentration reduction almost of 2-folds at 100°C (3.06 mg/100 g). It has been reported that carotenoids are heat sensitive components that

can be significantly deteriorated by high temperature treatments (Rodriguez-Amaya and Kimura 2004). In comparison to the TC values obtained in our study, Gopalan *et al.* (2004) and Sangwan *et al.* (2014) reported very low concentrations of carotenoids ranging from 0.68 to 0.81 mg/100 g in the ginger powder prepared by different drying methods such as shade, solar, oven and microwave. The obtained differences may be due to the effect of the applied treatment on the ginger, the sensitivity of the carotenoids and the varietal difference in the ginger selected for the study.

Analytical characteristics by HPLC

6-Gingerol, 6-shogaol and zingerone compounds

Ginger contains a number of important phenolic and pharmacologically active compounds particularly gingerols and shogaols. The pungency of fresh ginger is attributed to the presence of gingerols with 6-gingerol compound (1-[4'-hydroxy-3'-methoxyphenyl]-5'-hydroxy-3-decanone) as the major compound. The identification and quantification of the 6-gingerol, 6-shogaol and zingerone components were performed by HPLC assay as described in section 2. 8, and a representative typical chromatogram is shown in Figure 2 and the mean values are given in Table 3.

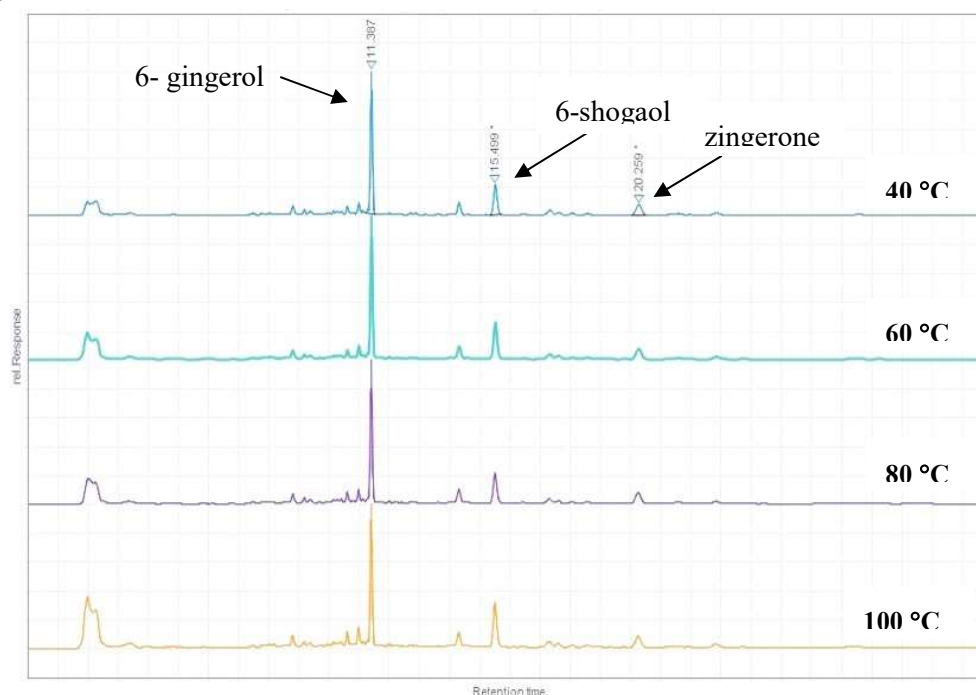


Figure 2. HPLC chromatograms of 6-gingerol, 6-shogaol and zingerone of the four ginger powders obtained at different drying temperatures

The results clearly showed that these components in the dried gingers were significantly ($P < 0.05$) influenced by the drying temperatures. The amount of 6-

gingerol gradually decreased as the applied drying temperature increased. As it is shown in Table 3, the value of 6-gingerol decreased from 4.96 mg/g at 40°C to reach the lowest value of 2.84 mg/g at 100°C. This finding is in agreement to that of Bhattarai *et al.* (2001) and Li *et al.* (2016). Gingerols are thermally labile molecules. During a drying process, a higher applied temperature would promote the decomposition and transformation of gingerols (6- gingerol, 8- gingerol and 10-gingerol) to shogaols (6-shogaol, 8-shogoal and 10-shagoal) (Huang *et al.*, 2011; Puengphian and Sirichote, 2008). Consequently, the amount of 6- gingerol decreased as the drying temperature increased. Additionally, it should be noted that different 6-gingerol contents have been reported by other studies in dried ginger powders. An *et al.* (2016) reported a lower value (2.50 mg/g) of 6-gingerol in the ginger powder obtained at 60°C compared to the relevant result obtained in this study, whereas, Li *et al.* (2016) reported a higher content (5.7 mg/g) of 6-gingerol in the ginger samples dried at 40°C. These findings highlight the sensitivity of 6-gingerol to high temperatures which leads to its transformation to shogaols. With respect to the shogaols compounds, which are the dehydrated form of gingerols, the increasing of the drying temperature over 60°C led to a significant ($P<0.05$) increase for the 6-shogaol compound that reached the highest content (5.82 mg/g) under the 100°C treatment (Table 3). In agreement to these findings, Huang *et al.* (2011) reported that the 80°C hot air drying led to a higher conversion rate of 6-shogaol compared to that obtained at 40°C. Moreover, Cheng *et al.* (2011) and Li *et al.* (2016) reported a similar trend in the 6-shogaol conversion rate and concluded that the transformation process of gingerols to shogaols is conditioned by the employed drying temperature and the drying time.

Table 3. Effect of drying temperatures on 6-gingerol, 6-shogoal, Zingerone and β -carotene compounds obtained by HPLC assay

Treatment	6-gingerol, mg/g	6-shogaol, mg/g	Zingerone, mg/g	β -carotene, mg/100g
40°C	4.96 ± 0.17 ^a	4.88 ± 0.18 ^c	0.20 ± 0.04 ^b	2.56 ± 0.23 ^a
60°C	4.31 ± 0.21 ^b	4.96 ± 0.14 ^c	0.22 ± 0.06 ^{ab}	2.36 ± 0.42 ^{ab}
80°C	4.08 ± 0.10 ^c	5.43 ± 0.16 ^b	0.23 ± 0.04 ^{ab}	2.02 ± 0.81 ^{ab}
100°C	2.84 ± 0.04 ^d	5.82 ± 0.20 ^a	0.25 ± 0.01 ^a	1.79 ± 0.52 ^b

The values presented are the means ± standard deviation (n=3). The means in the same column bearing a different superscript letter differ significantly ($P<0.05$)

Like shogaols, zingerone (4-(4-hydroxy-3-methoxyphenyl)-2-butanone) is another compound that is absent in fresh ginger and generated during the processing of ginger at high temperature. As it can be seen from the data in Table 3, the zingerone content was positively related to the applied drying temperature. The drying process at 100°C exhibited a significantly higher ($P<0.05$) value of zingerone (0.25 mg/g) compared to the one obtained after the 40°C treatment (0.20 mg/g). In this context, Li *et al.* (2016) reported a similar trend for the dried ginger samples and concluded that zingerone concentrations are enhanced by high

processing temperatures. However, these authors reported a lower content of zingerone (0.158 mg/g) compared to our results.

β-carotene compound

A comparison between the β -carotene contents as obtained by the HPLC assay after different ginger treatments is presented in Table 3. The results revealed that β -carotene values were inversely related to the applied drying temperature. The ginger powder dried at 40°C showed significantly ($P<0.05$) the highest β -carotene content (2.56 mg/100 g), which decreased constantly by rising the drying temperature so as to at 100°C it reached the lowest value (1.79 mg/100 g). These results are in accordance to the total carotenoids (section 3.2.3, Table 2) and are also in agreement to those of Georgé *et al.* (2011), Arkoub-Djermoune *et al.* (2016) and Lu *et al.* (2018) who elucidated the inverse relationship between β -carotene and high temperature processing, light and storage period. These factors may cause the isomerisation and oxidation of β -carotene, followed by a cleavage due to its unsaturated structure.

Compositional properties of the ginger powders

The mean values of moisture, total protein, water-soluble proteins and fat contents determined by rapid techniques (see section 2.10) are given in Table 4. The moisture content of the ginger powders significantly ($P<0.05$) decreased once the drying temperatures were increased. In this context, Shirin Adel and Prakash (2010) and Sangwan *et al.* (2014) reported moisture values of 6.67% and 15.02% for the ginger powder obtained by drying at 40° and 50°C, respectively.

Contrary to the moisture result trend, the fat content was positively related to the drying process temperatures. This result could be attributed to the increased total solids in the ginger powders by increasing the drying temperatures. It should be noted that the difference between the highest and the lowest obtained fat values were 0.18%, though the differences were significant ($P<0.05$). In comparison to our results, Shirin Adel and Prakash (2010) and Odebumni *et al.* (2010) reported a higher fat content for the ginger powder dried at 40°C (3.72- and 5.62%, respectively), while Sangwan *et al.* (2014) found a less fat content (0.78%) at 50°C.

Table 4. Chemical composition of the ginger extracts obtained at different drying temperatures

Treatment	Moisture (%)	TPr (%)	WSP (%)	Fat (%)
40°C	8.21 ± 0.09 ^a	8.37 ± 0.13 ^a	1.17 ± 0.03 ^a	1.20 ± 0.06 ^a
60°C	7.28 ± 0.09 ^b	8.45 ± 0.11 ^b	1.15 ± 0.02 ^a	1.24 ± 0.07 ^b
80°C	6.46 ± 0.16 ^c	8.77 ± 0.14 ^c	1.12 ± 0.01 ^b	1.29 ± 0.05 ^{bc}
100°C	4.58 ± 0.10 ^d	9.08 ± 0.12 ^c	1.07 ± 0.03 ^c	1.38 ± 0.10 ^c

The values presented are the means ± standard deviation (n=3). The means in the same column bearing a different superscript letter differ significantly ($P<0.05$). TPr: Total protein; WSP: Water-soluble protein.

With respect to the protein determination, the total protein (TPr %) content increased significantly ($P<0.05$) in the ginger powders by increasing the drying

process temperatures. The ginger powder obtained after the drying process at 100°C exhibited the highest TPr value (9.08%). This value was steadily and significantly decreased by decreasing the drying temperatures to reach the value of 8.37% TPr in the ginger powder obtained after the 40°C treatment (Table 4). Other studies on oven dried ginger reported a lower TPr values compared to the results obtained in this study. Shirin Adel and Prakash (2010) found a 5.09% protein content in the ginger powder dried at 40°C, while Sangwan *et al.* (2014) reported a total protein content of 5.70% in the ginger dried at 50°C. The level of water-soluble proteins (WSP) was determined as an indicator for the impact of the drying temperatures on the availability of small and soluble proteins in the ginger powder.

Contrary to the TPr, the WSP contents were inversely related to the applied drying temperatures. The highest amount (1.17 %) was obtained at 40°C. This amount steadily decreased to reach the lowest value (1.07 %) at 100°C. This decrease is most likely attributed to the denaturation and changes in the molecular and structural characteristics of the small proteins by increasing the drying temperature that caused more proteins to bind and to become insoluble in water. With respect to the statistical analysis, although the differences among the WSP values were very small, they appeared significantly different ($P < 0.05$) most likely due to the very low values of the adjusted mean square error obtained in comparison to the means (0.0004). Overall, it is worth noting that the differences in the chemical composition of the ginger powders obtained in this study compared to other studies could be attributed to the geographic origin, the atmospheric conditions and the maturity of the used rhizomes.

Antioxidant activity

To evaluate the influence of different drying temperatures on the antioxidant activity of ethanol ginger extracts, the DPPH radical scavenging assay was performed and the results are presented in Table 2. The results revealed that the inhibition percentage of the DPPH radical was significantly increased ($P < 0.05$) in ginger by increasing the drying temperatures. The ginger sample dried at 100°C showed the highest DPPH scavenging activity equivalent to 78.23%, whereas the minimum DPPH scavenging activity value corresponding to 73.47% was obtained at 40°C. It has been reported that phenolic compounds are potential antioxidants and free radical-scavengers and a good correlation should be obtained between the content of phenolic compound and the antioxidant activity (Kumar *et al.*, 2008; An *et al.*, 2016). This result is in accordance with the result of total phenolic and total flavonoid contents of the same treatments (Table 2), thus reflecting the higher antioxidant activity obtained by increasing the drying temperature. In agreement to these findings, Praven *et al.* (2007), and Hasna and Afidah (2009) concluded that the ginger with high levels of TP and TF contents exhibited a high free radical scavenging activity. Furthermore, the obtained DPPH scavenging ability showed a significant ($P < 0.01$) positive correlations with the TP and TF contents (0.89 and 0.92, respectively). Similar trend of the correlations was reported by Velioglu *et al.* (1998), and Shirin Adel and Prakash (2010). In addition, a significant ($P < 0.01$) inverse correlation was obtained between the radical scavenging activity and TC

content (-0.96). This correlation would be expected, since TC values showed a significant reduction trend by increasing the drying temperatures and were in contrast to TP and TF contents (Table 2). Accordingly, these findings may suggest that the high antioxidant activity of ginger extracts may be attributed to the phenolic compounds including flavonoids and other released compounds that are not considered in the current study, rather than carotenoids.

Microstructure of ginger powders

The effects of the applied oven drying temperatures on the microstructure characteristics of the ginger powder are shown in Figure 3. The irregularly aggregated parenchyma tissue formed the main matrix (pointed with black arrows), and the scattered starch grains (pointed with white arrows) throughout the parenchyma made the structural dimensions of these images. The SEM micrographs clearly showed the destruction of the parenchyma cell wall and the absence of the oils cells in the ginger powders as a result of the drying process. The same observations were reported by Azian *et al.* (2004) and An *et al.* (2016). Fig. 3A and 3B which represented the ginger powders dried at 40°C and 60°C showed abundant starch grains of well-retained shape that were dispersed all over the parenchyma matrix and were more pronounced compared to the ginger's structure after the 100°C treatment (Figure 3D). It should be noted that the starch grains of the ginger dried at 80°C (Figure 3C) were markedly destroyed and embedded within the parenchyma matrix compared to the 40°C and 60°C treatments.

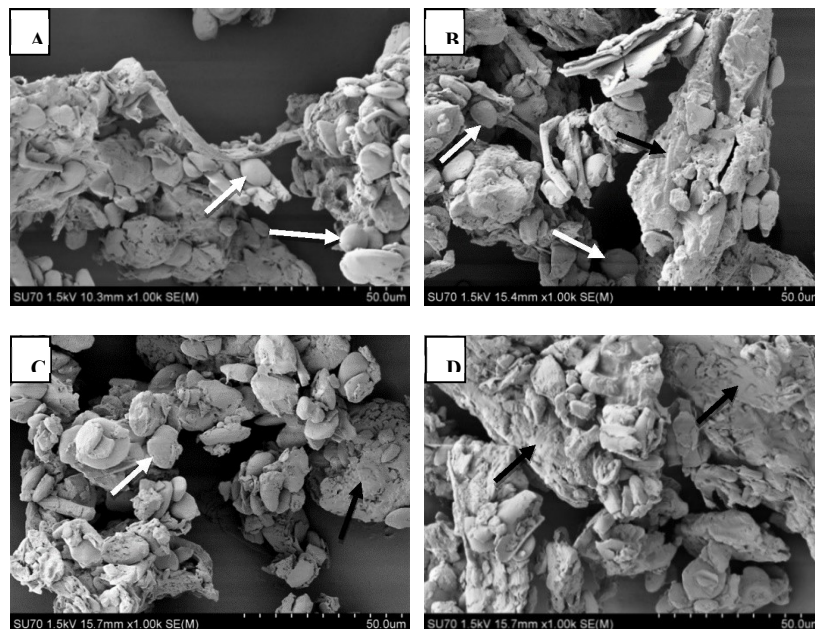


Figure 3. SEM micrographs (1000x) of the ginger powders obtained at four different drying temperatures. (A) 40°C; (B) 60°C; (C) 80°C and (D) 100°C. *Black arrows* indicate the ginger parenchyma tissue aggregates while the *white arrows* indicate the starch grains. Scale bar is 50 μm

Additionally, the microstructure of the ginger powder obtained after the 100°C drying process, as shown in Figure 3D, revealed a denser structure, with a more pronounced degree of gelatinization and a higher destruction degree of the starch granules compared to those of the ginger powders obtained at lower drying temperatures 40°, 60°, and 80°C (Figures 3A, 3B and 3C, respectively). This observation is in agreement to the one of Huang *et al.* (2011) who reported a serious broken down phenomenon of the ginger starch dried at 80°C. In this context, An *et al.* (2016) reported that a high temperature and a rapid drying process make the inner water difficult to remove. This fact led the cellular structures to crosslink together so that the starch grain showed an important degree of gelatinization.

Conclusions

The present study provided some comparative results on the chemical and bioactivity characteristics of the ginger powders obtained through oven drying process using four different temperatures (40°, 60°, 80° and 100°C). The results led to the conclusion that using an elevated temperature is beneficial in the drying process due to certain associated advantages. For instance, it exhibited a positive effect on the preservation of the total polyphenols, total flavonoids, and consequently higher antioxidant activity. Zingerone, 6-shogaol, total protein and fat contents of the ginger powders were significantly and positively related to the applied drying temperatures. Whereas, 6-gingerol and total carotenoids were inversely related to the drying temperatures, most probably due to the degradation of these compounds caused by high drying temperatures. Accordingly, the 100°C treatment revealed the most appropriate drying temperature in terms of ginger powder quality attributes. Overall, this study emphasized the profound influence of both of the drying time and the drying temperature on the final composition of the ginger powder. The obtained results offer a better comprehension of the influence of the oven drying process on the ginger powder functional properties and structure characteristics, allowing the optimization and the development of food and pharmaceutical industries applications.

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